

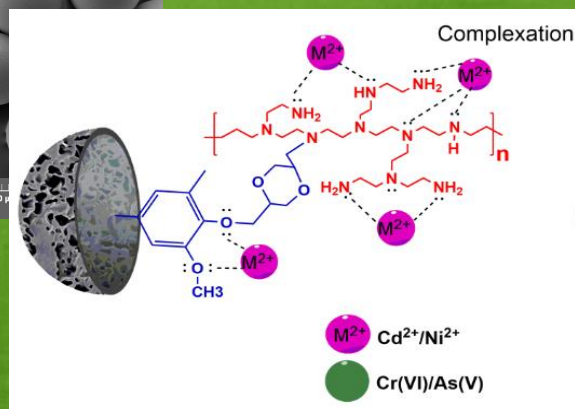
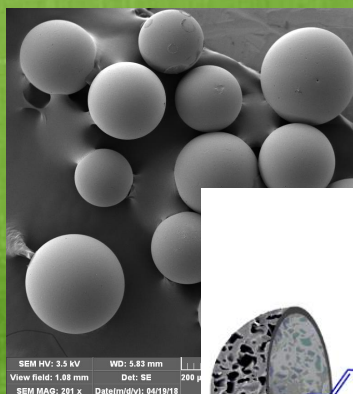


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Lignin microspheres - a novel eco-friendly adsorption material

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Introduction: Why lignin?

- **waste by-product** from the pulp and paper industry or biomass pretreatment processes
- **underexploited**
- possibility of **valorization**
- second generation biomass
- substance with a **binding** properties that causes the **compactness** of wood cell structure
- **abundant**, low-in cost, and **eco-friendly**
- **kraft lignin** was used in our research



UNF Lignin-based microspheres

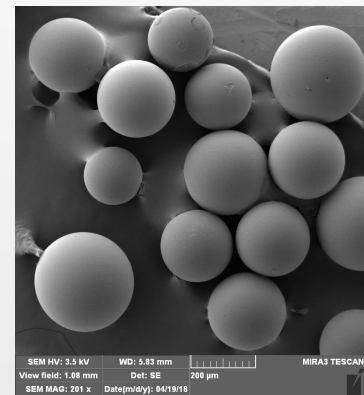
- The presence of various **functional groups** (aliphatic and aromatic/phenolic OH - groups) makes lignin a suitable substance for chemical syntheses to form **polymer materials**

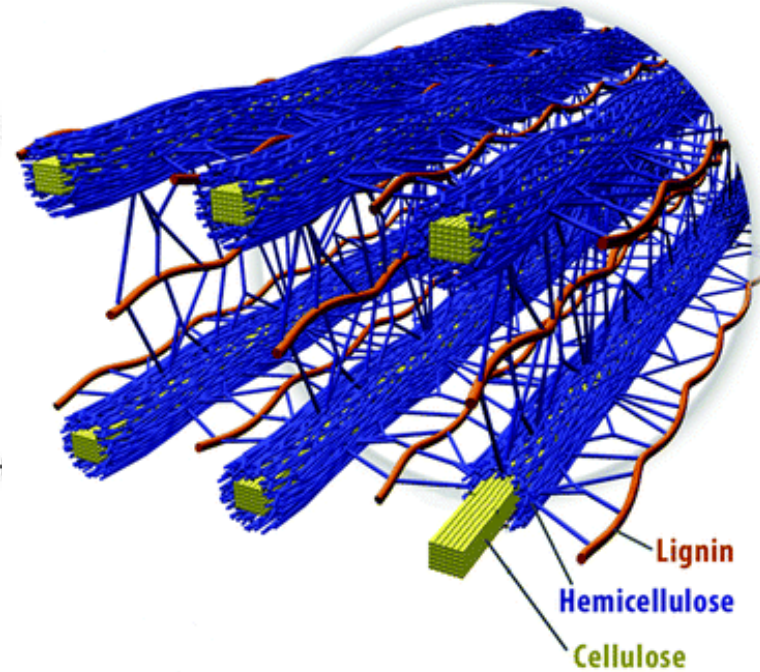
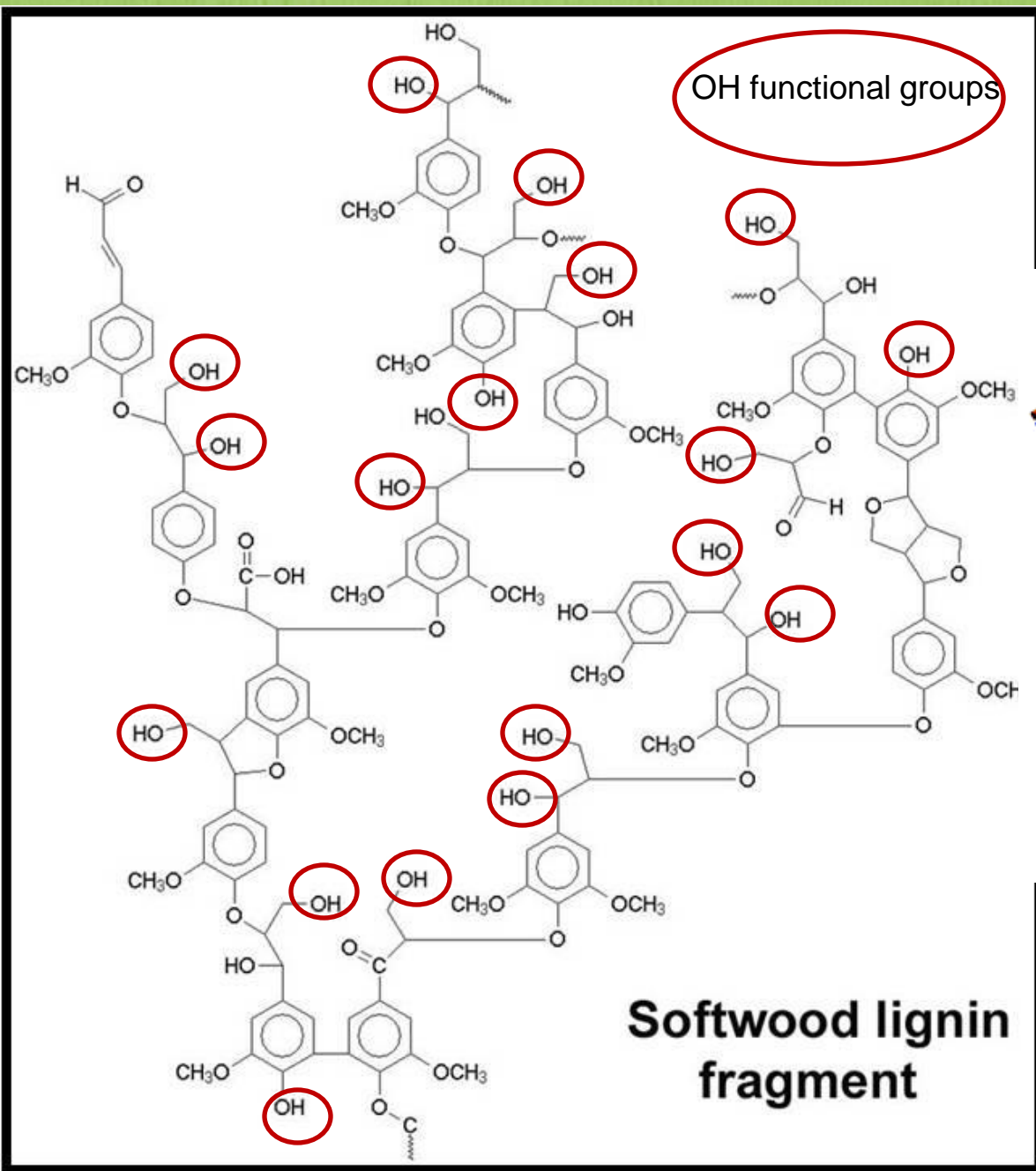
Our research product:

- **Lignin microspheres (LMS)**: An effective and recyclable natural polymer-based **adsorbent** for **diclofenac** and **heavy-metal ion** removal

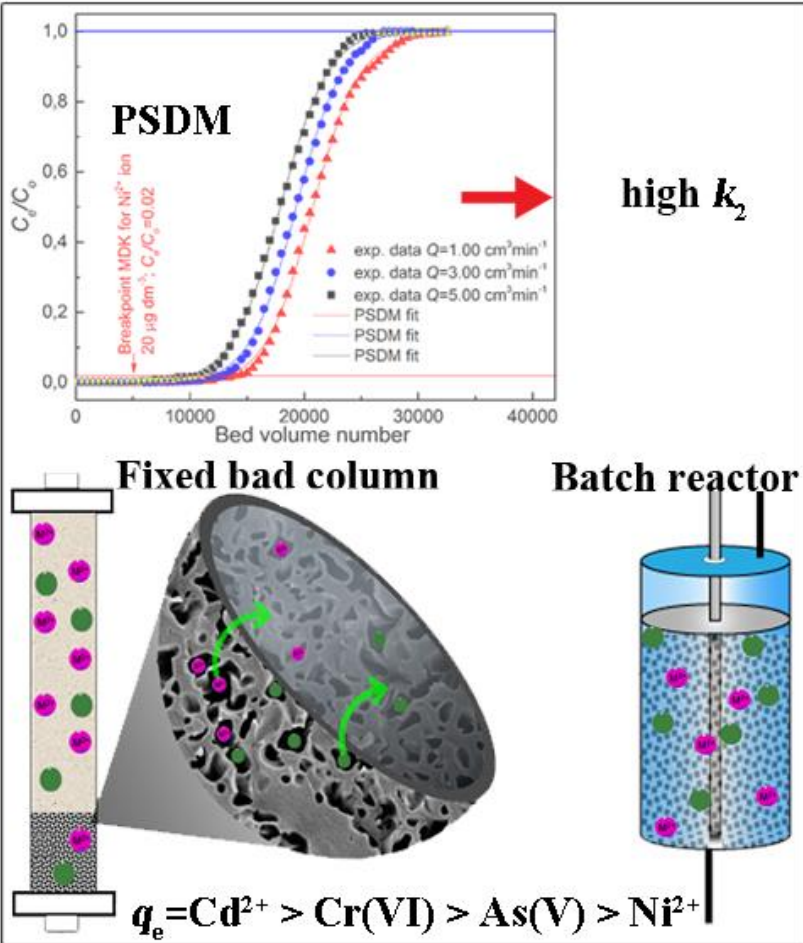
2 methods of synthesis of LMS we used:

- inverse suspension copolymerization of lignin
- copolymerization of lignin methacrylic derivatives prepared with methacryloyl chloride





Ge et al. 2017



Results

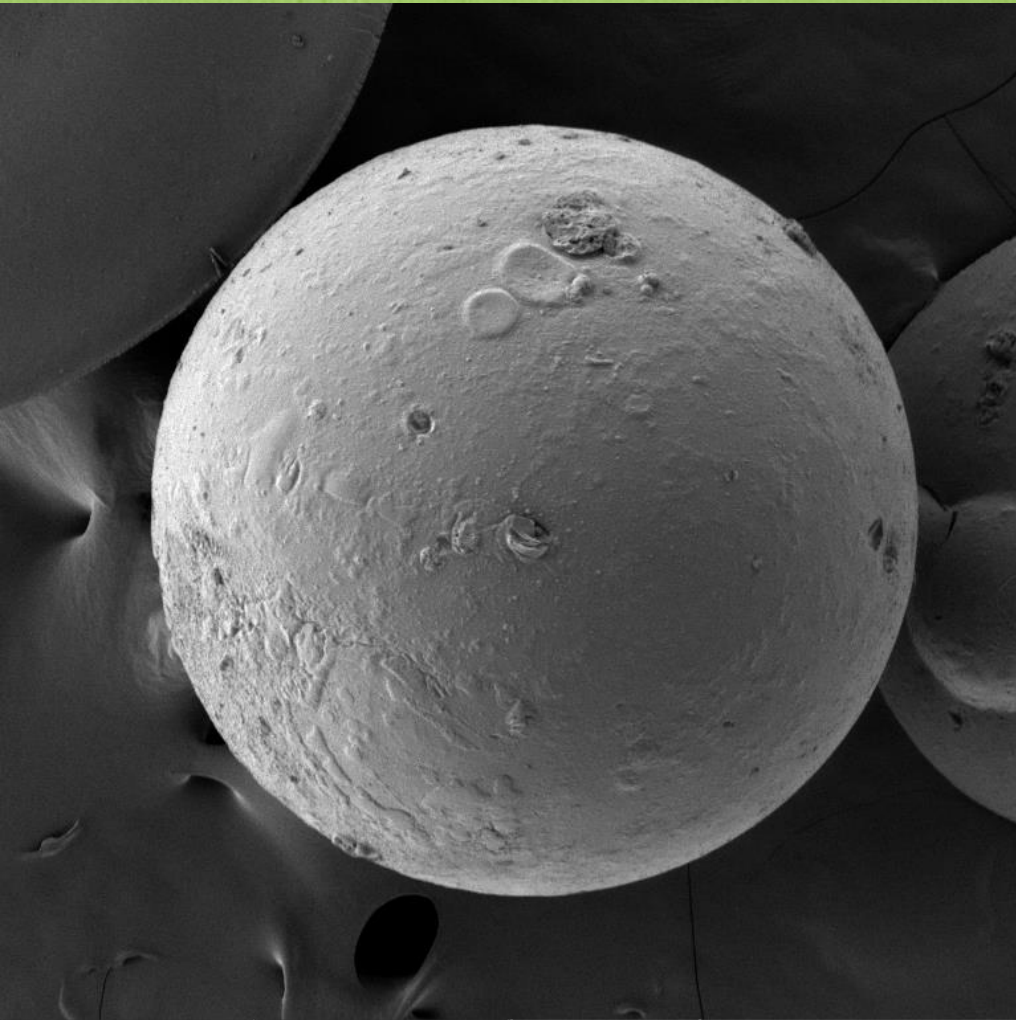
- 1 (LMS-1%): LMS, with 1.0 wt.% Na-alginate solution
- 2 (LMS-5%): LMS, with 5.0 wt.% Na-alginate solution
- 3 (LMS-10%): LMS, with 10.0 wt.% Na-alginate solution
- 4 (LMS-Fe₃O₄): LMS, with amino-modified Fe₃O₄ nanoparticles
- 5 (LMS-MnO₂): LMS, with amino-modified MnO₂ nanoparticles
- 6 (L-MAC-MS-TMPTA): Lignin methacrylic deriv. MS, TMPTA*
- 7 (L-MAC-MS-TEGDMA): Lignin methacrylic deriv. MS, TEGDMA*
- 8 (L-MAC-MS-MEMO-Fe₃O₄): LMS with MEMO* modified Fe₃O₄
- 9 (L-MAC-MS-MAC-Fe₃O₄): LMS with methacrylic modified Fe₃O₄
- 10 (L-MAC-MS-MAC-MnO₂): LMS with methacrylic modified MnO₂

* Trimethylolpropane triacrylate; * Triethylene glycol dimethacrylate; * Methacrylfunctional silane

Characterisation

- The course of the modification of lignin was confirmed by **FTIR**, nuclear magnetic resonance (**NMR**) spectroscopy and X-ray photoelectron spectroscopy (**XPS**)
- The porous structures and morphology of obtained LMS were investigated by scanning electron microscopy (**SEM**), as well as via Brunauer-Emmett-Teller surface area analysis (**BET**) and Barrett-Joyner-Halenda (**BJH**) pore size and volume analysis
- X-ray diffraction analysis (**XRD**), thermo-gravimetric analysis (**TGA**)
- Amino-group content determined *via* "back" (**indirect**) titration
- Transmission electron microscopy (**TEM**) for analysis of MnO₂ and Fe₃O₄ nanoparticles
- **Adsorption** experiments (**batch** and **column** studies) were made with diclofenac (DCF), *Ni(II)*, *Cd(II)*, *arsenate* and *chromate ions*

SEM view

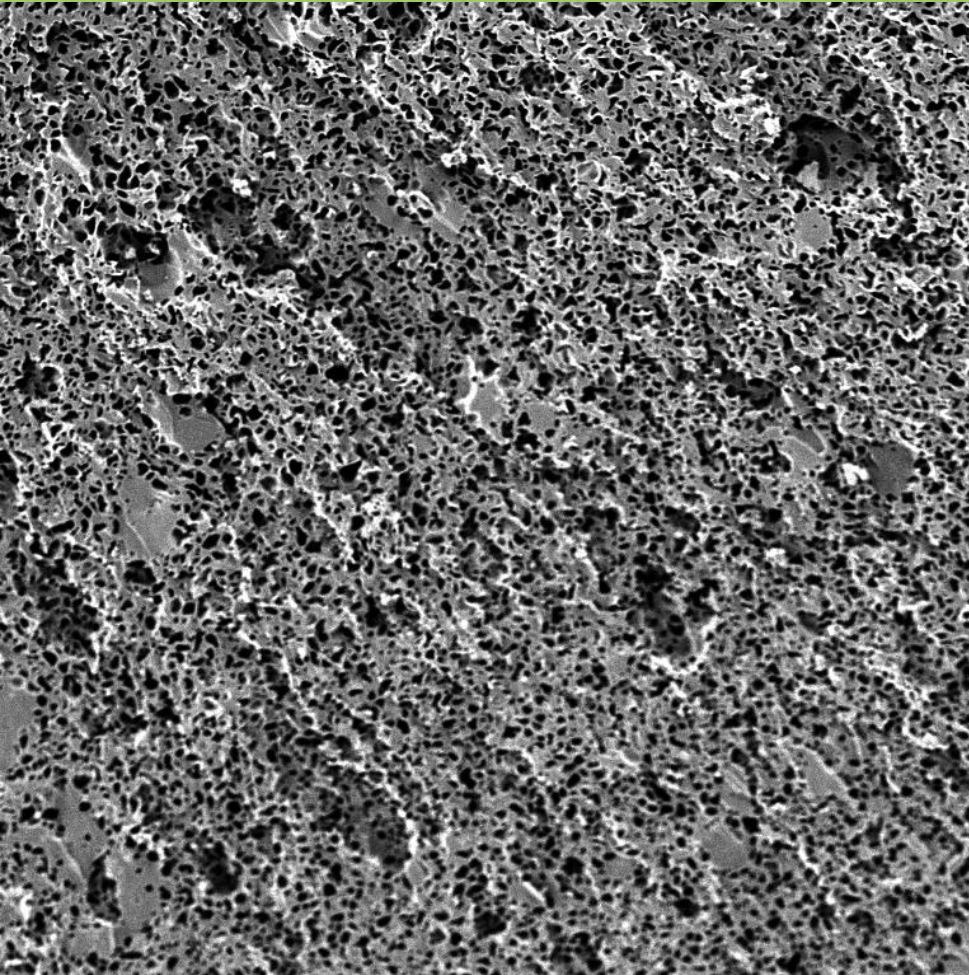


SEM HV: 3.5 kV	WD: 5.25 mm		MIRA3 TESCAN
View field: 1.08 mm	Det: SE	200 μm	
SEM MAG: 200 x	Date(m/d/y): 04/19/18		

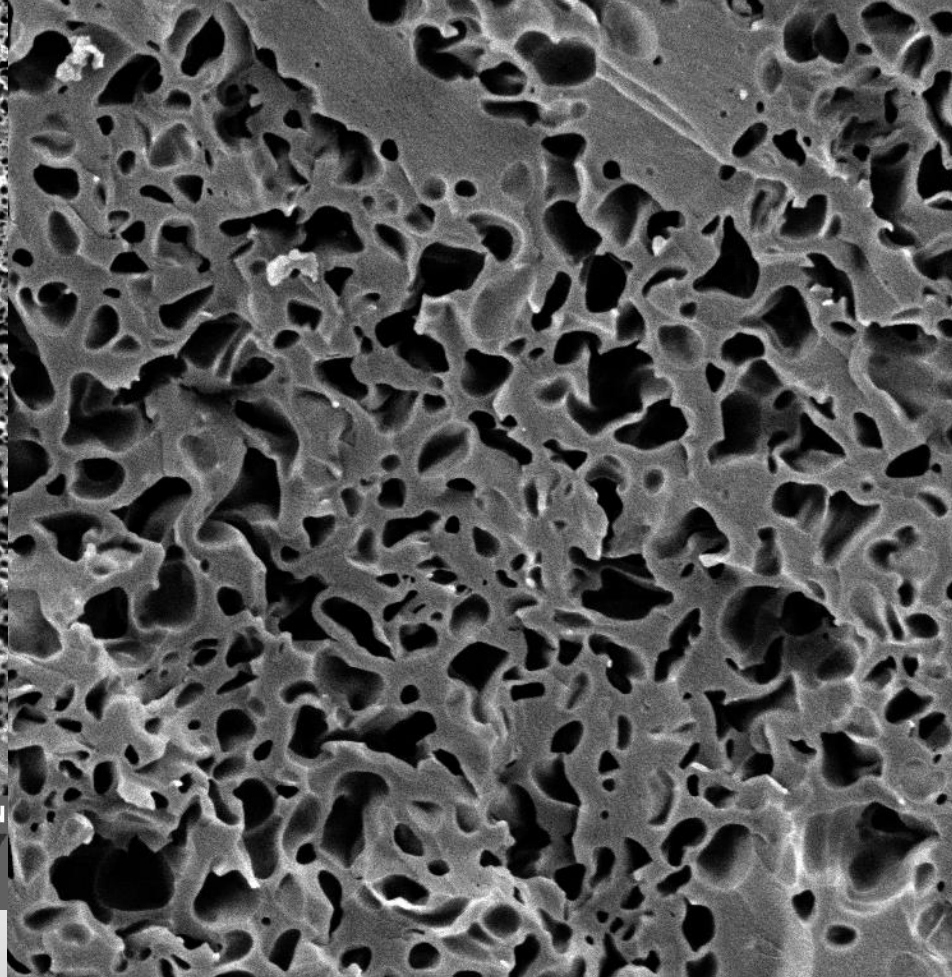


SEM HV: 3.5 kV	WD: 5.76 mm		MIRA3 TESCAN
View field: 578 μm	Det: SE	100 μm	
SEM MAG: 375 x	Date(m/d/y): 04/19/18		

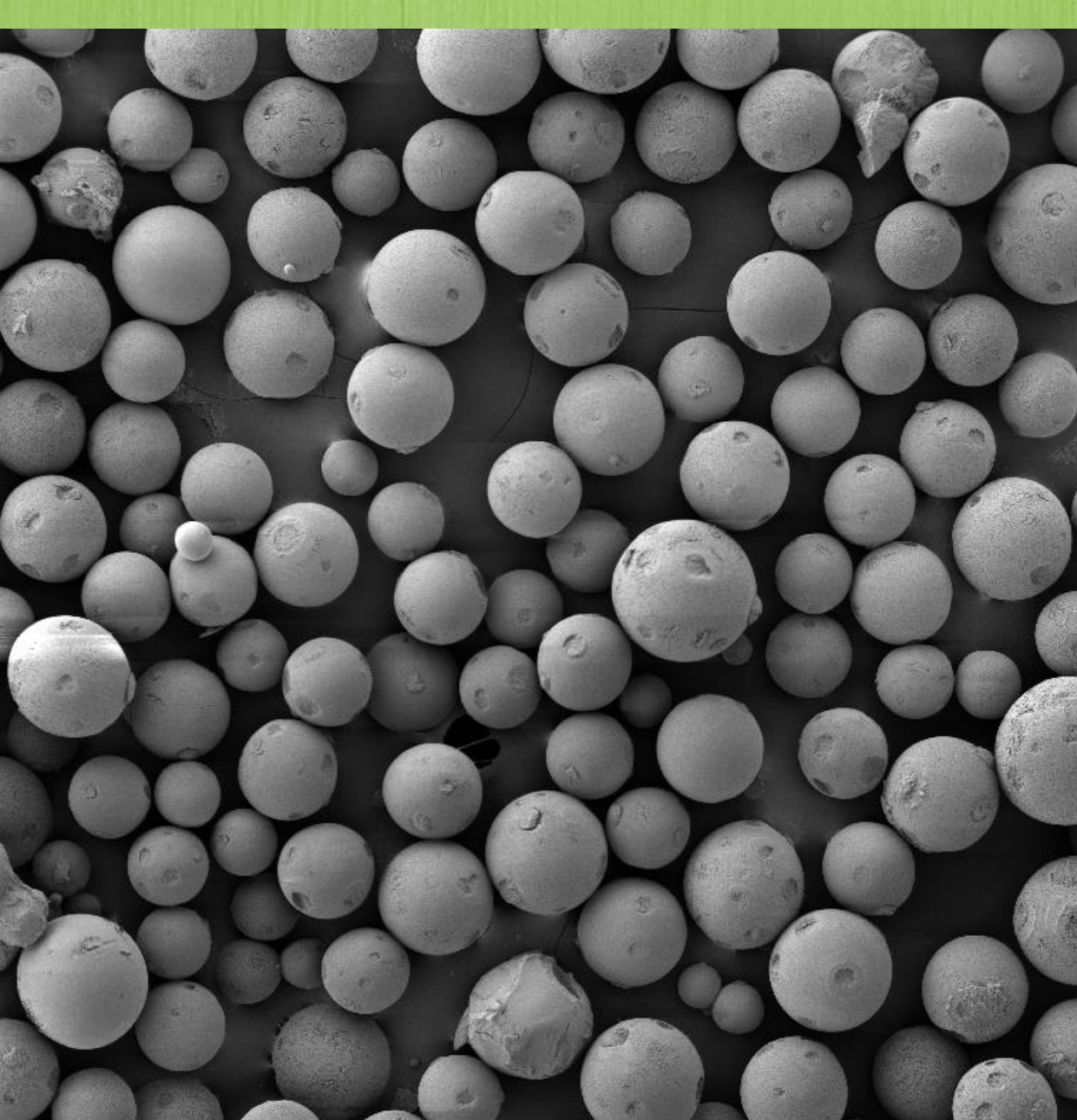
High porosity



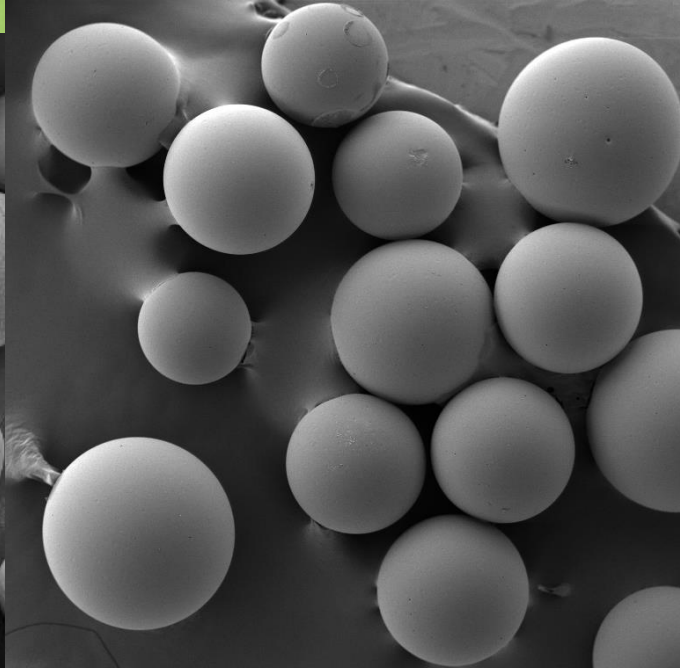
SEM HV: 3.5 kV	WD: 5.59 mm		MIRA3 TESCAN
View field: 108 μm	Det: SE	20 μm	
SEM MAG: 2.00 kx	Date(m/d/y): 04/19/18		



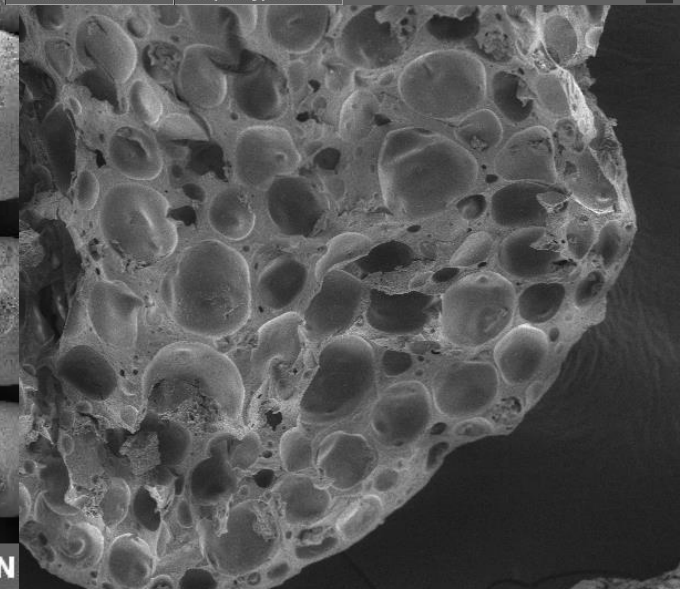
SEM HV: 3.5 kV	WD: 5.62 mm		MIRA3 TESCAN
View field: 21.7 μm	Det: SE	5 μm	
SEM MAG: 10.00 kx	Date(m/d/y): 04/19/18		



SEM HV: 3.5 kV	WD: 6.10 mm		MIRA3 TESCAN
View field: 1.08 mm	Det: SE	200 µm	
SEM MAG: 200 x	Date(m/d/y): 04/19/18		

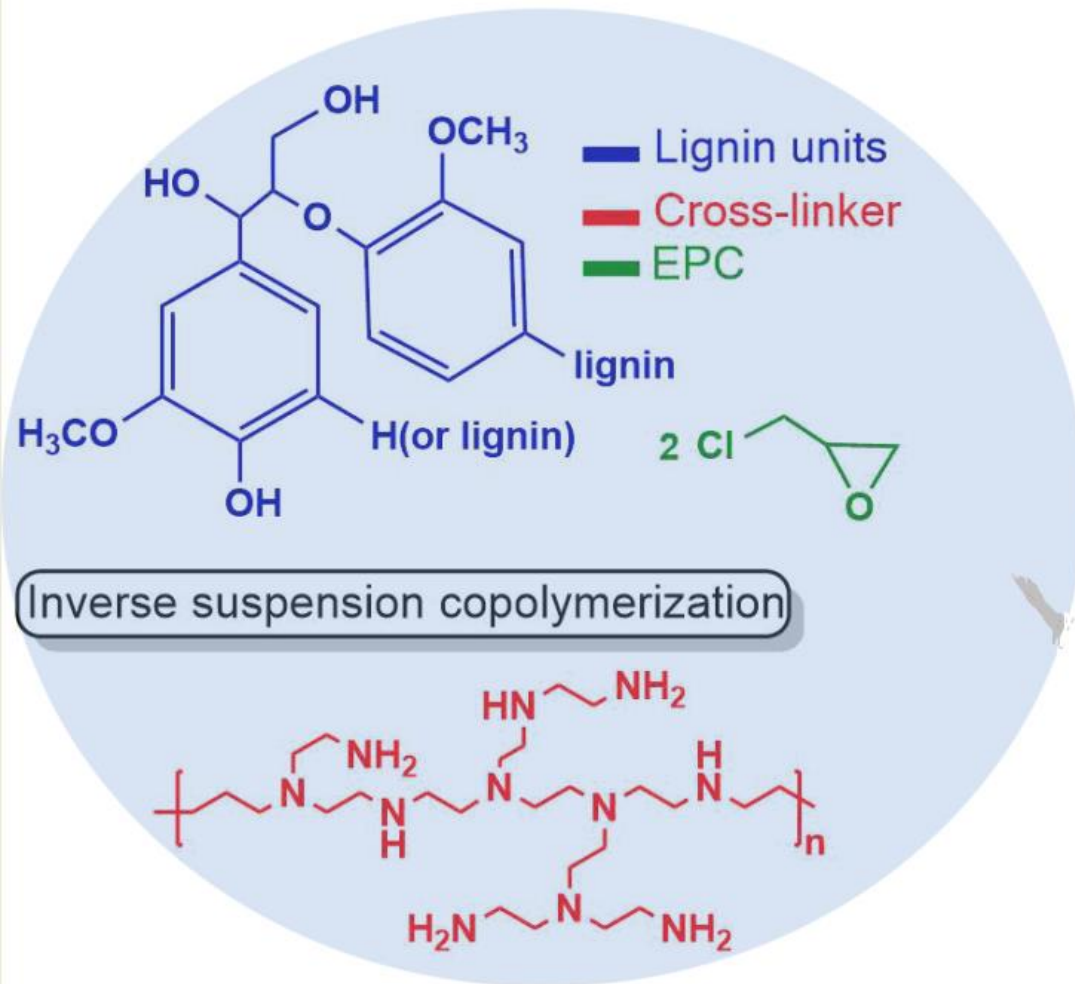


SEM HV: 3.5 kV	WD: 5.83 mm		MIRA3 TESCAN
View field: 1.08 mm	Det: SE	200 µm	
SEM MAG: 201 x	Date(m/d/y): 04/19/18		



WD: 5.92 mm		MIRA3 TESCAN
Det: SE	100 µm	
Date(m/d/y): 04/19/18		

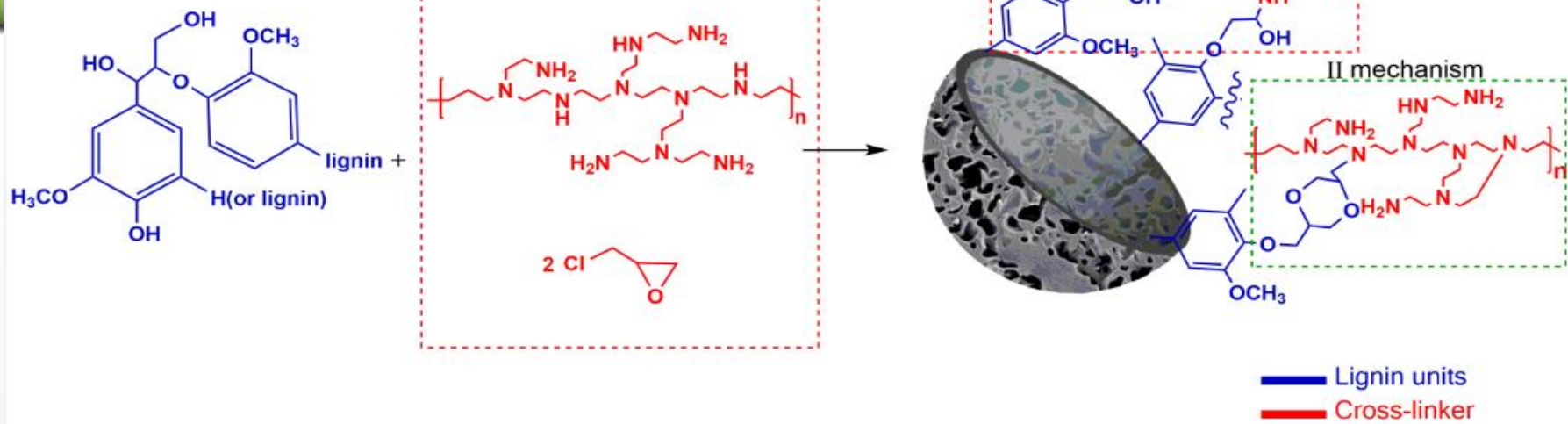
Inverse suspension copolymerization



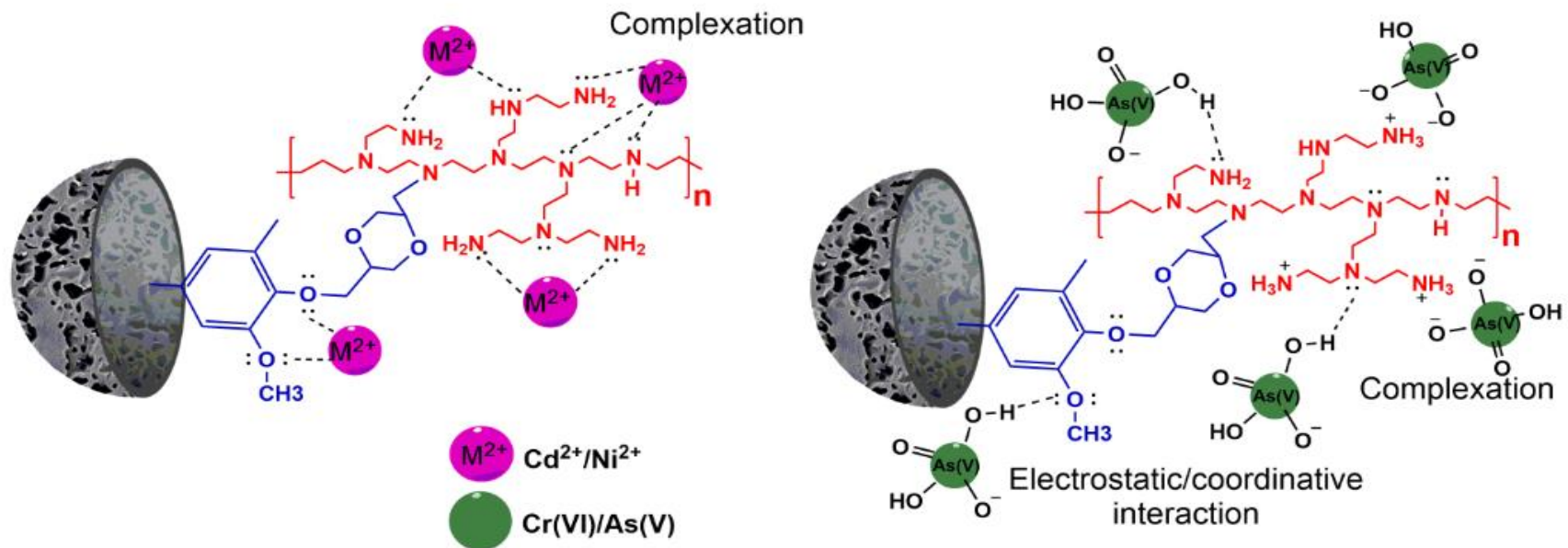
A-LMS SYNTHESIS

- kraft lignin
- poly(ethylene imine) grafting-agent
 - for amino-functionalization
- epoxy chloropropane cross-linker
- sodium alginate emulsifier (1, 5 and 10 wt.%)

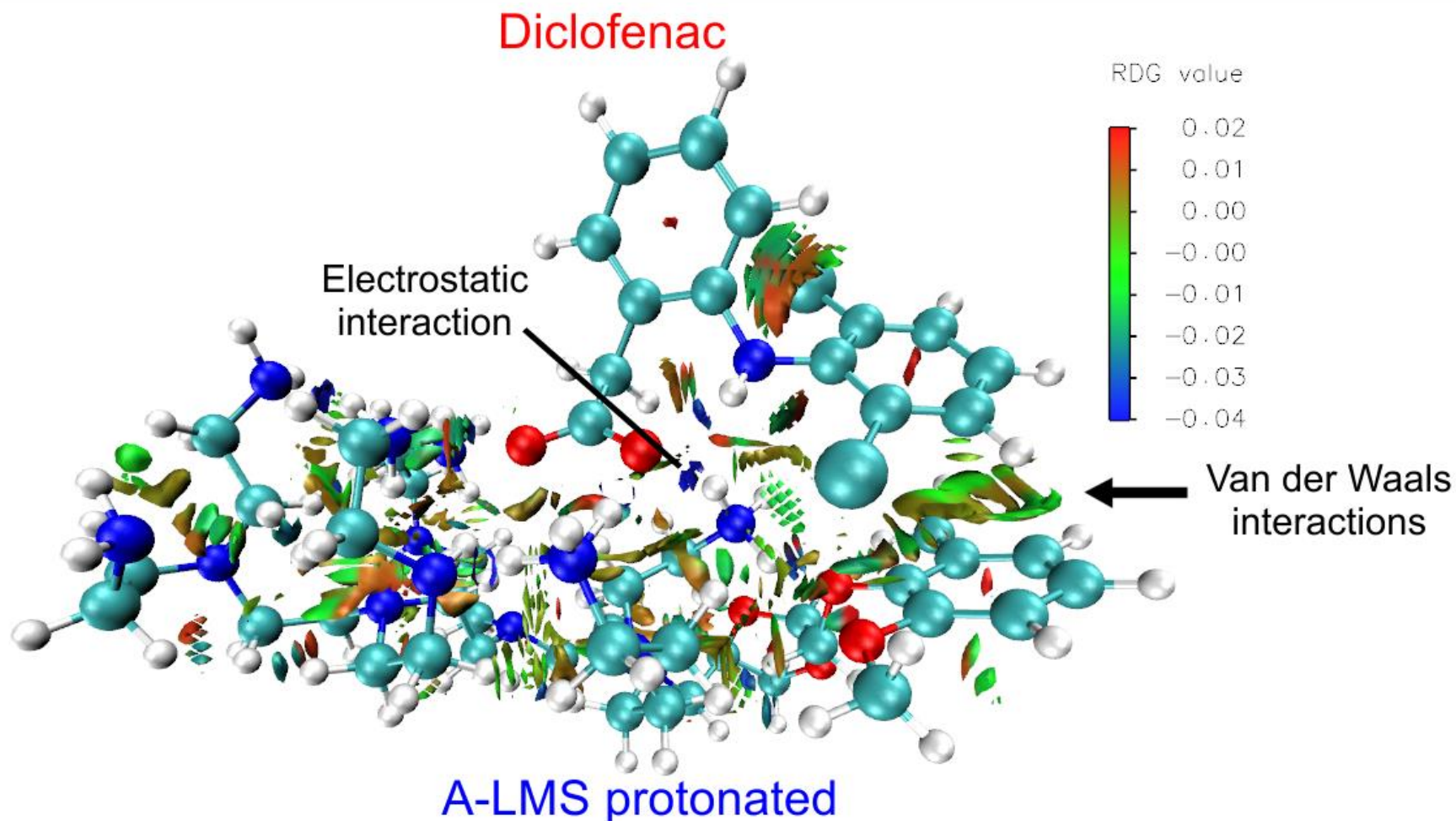
a.) Exploring mechanisms of synthesis



b.) Exploring mechanisms of heavy-metal ion adsorption



Exploring the diclofenac adsorption



Adsorption isotherms, kinetics and column study

- Langmuir, Freundlich and Dubinin-Radushkevich adsorption isotherm fitting data, obtained at 298, 308 and 318 K
- Thermodynamic parameters of adsorption analyzed
- Kinetic study was done with the following kinetic models:
 - PSO,
 - Roginsky-Zeldovich-Elovich (Elovich)
 - Dunwald-Wagner (DW)
 - Homogenous Solid Diffusion (HSDM), as well as
 - Intra-particle **Weber-Morris** (W-M) Model
- Fixed-bed column studies using Bohart-Adams, Yoon-Nelson, Thomas and Clark models
- Pore surface diffusion modeling (PSDM) – predicting the full-scale packed system

Conslusions

- The proposed LMS are **eco-friendly**, **cost-efficient** and **effective** adsorption materials
- Proven presence of **functional groups** and well-developed, highly **porous structure**
- LMS showed an **excellent adsorption capacity** towards *diclofenac* and heavy-metal ions: *Ni(II)*, *Cd(II)*, *arsenate* and *chromate ions*
- DCF (**151,13**) >> Cd²⁺ (74,84) > Cr(VI) (54,20) > As(V) (53,12) > Ni²⁺ (49,42 mg g⁻¹)
- Formulation of lignin-based **new bio-polymer materials** increase the contribution of **renewable** materials, **from waste resources**, in wastewater treatment processes

Thank you!



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Novel amino-functionalized lignin microspheres: High performance biosorbent with enhanced capacity for heavy metal ion removal

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Kinetics and column adsorption study of diclofenac and heavy-metal ions removal by amino-functionalized lignin microspheres

Ana L. Popovic ^a, Jelena D. Rusmirovic ^{b, c}, Zlate Velickovic ^c, Tihomir Kovacevic ^b, Aleksandar Jovanovic ^a, Ilija Cvijetic ^d, Aleksandar D. Marinkovic ^a

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